A Simple and Sensitive One-Dimensional NMR Technique for Correlation of Proton and Carbon Chemical Shifts

SUSANTA K. SARKAR AND AD BAX

Laboratory of Chemical Physics, National Institute of Arthritis, Diabetes and Digestive and Kidney Diseases, National Institutes of Health, Bethesda, Maryland 20205

Received June 4, 1984; revised October 23, 1984

A slightly modified version of the well-known selective population transfer (SPT) experiment (1-5) is described that is very easy to use and correlates ¹H and ¹³C or ¹⁵N chemical shifts with high sensitivity and accuracy. Although the changes with the original experiment (1) are relatively small, they are helpful for reliable and convenient use of the experiment.

Two-dimensional heteronuclear chemical-shift correlation (6-10) has proven to be a powerful tool for the assignment of ¹³C spectra. However, in practice, sensitivity of the two-dimensional experiment is at best about a factor of three lower (10) than that of a regular proton-decoupled ¹³C spectrum. Moreover, large data matrices are usually required in this type of 2D experiment and minimum data acquisition plus processing time often exceeds one hour.

Recently, there has been a renewed interest in alternative one-dimensional experiments that provide essentially the same information as the 2D experiment by transfer of magnetization from a preselected proton to its coupled ¹³C nucleus (11-14), or by suppression of the resonance of a ¹³C that is coupled to a preselected proton (15). Those sequences are all very elegant, but are for most purposes unnecessarily complicated and suffer more from relaxation effects than necessary. Furthermore, in some of those experiments confusion may arise about whether the magnetization has been transferred from the high-field carbon satellite of one proton or from the low-field satellite of another proton that resonates at a frequency J_{CH} lower than the first proton.

In the SPT experiment (Fig. 1), a soft ¹H 180° pulse ($\gamma H_2/2\pi = 20$ Hz) selectively inverts proton resonances of either the low-field or the high-field ¹³C satellite. For a methine group, this changes the populations of the energy levels in such a way that the two ¹³C doublet components change in intensity by a factor of ±4 compared with their thermal equilibrium values (without NOE). The theory of this phenomenon has been described in the original literature (1-5), and will not be repeated here. Broadband proton decoupling cannot be started immediately after the ¹³C detection pulse, as this would cause mutual cancellation of the antiphase ¹³C doublet components. However, a delay $1/(2J_{CH})$ for methine groups, or $1/(4J_{CH})$ for methylene and methyl groups inserted before decoupling is started (Fig. 1) will prevent signal cancellation and allow proton-decoupled ¹³C spectra to be recorded



FIG. 1. Pulse sequence of the decoupled SPT experiment. Saturation of the ¹³C signal is obtained by the application of a number (>10) of 180° ¹³C pulses spaced by 2.2 times the width of the selective 180° ¹H pulse. The radiofrequency field strength ($\gamma H_2/2\pi$) of the soft proton 180° pulse is on the order of 20 Hz (pulse width 25 ms). Data acquisition is started immediately after the 90° observe pulse, and broadband decoupling is switched on a time, Δ , later. The duration of Δ is set to 1/(2J_{CH}) for methine resonances and to 1/(4J_{CH}) for methylene and methyl groups.

(16, 17). Note that in the case where the ^{13}C satellite of one of the two nonequivalent protons of a geminal pair is inverted, only a factor of two enhancement in the decoupled spectrum is obtained.

Because transverse ¹³C magnetization rotates during the time, Δ , under the influence of scalar coupling and chemical shift, a large frequency-dependent phase error of the decoupled SPT spectrum would be induced if acquisition is started at time, Δ , after the ¹³C detection pulse. The simplest way around this is to start data acquisition immediately after the observe pulse, while the decoupler is switched on at time, Δ (18). ¹³C signals due to SPT transfer will be either +90° or -90° out of phase relative to signals recorded in a conventional "one-pulse" ¹³C spectrum, depending on whether the high-field or the low-field ¹³C satellite has been inverted. A phase correction of +90° will phase those resonances to negative and positive absorption mode, respectively. A small baseline distortion will be induced by the fact that during the short time, Δ , no proton decoupling was applied. To suppress 13 C signals that are not due to SPT transfer, a number of 13 C 180° saturation pulses are applied prior to the selective 180° ¹H pulse (Fig. 1). Optimum suppression is obtained if the spacing between the 180° ¹³C pulses is about 2.2 times the duration of the selective ¹H 180° pulse. A suppression factor larger than 100 is easily obtained.

Figure 2 demonstrates the effectiveness of the method for a sample of chrysene, a potent carcinogen. A 16 mM solution in $CDCl_3$ in a 10 mm sample tube is used. The 500 MHz proton spectrum is easily assigned via a COSY experiment. The doublets of protons H-4 and H-5 are partially overlapping as their difference in chemical shift is only 0.016 ppm, and provide a challenging test for the accuracy of the SPT method. Figure 2b shows the regular NOE enhanced ¹³C spectrum of the protonated aromatic carbons, obtained from 200 accumulations (12 min). Aromatic CH couplings are approximately 160 Hz, and in the SPT experiments the selective proton chemical shifts. Figure 2c shows the SPT spectrum for transfer from the downfield satellite of H-3 and clearly shows a positive carbon resonance for C-3. As the selective proton pulse also affects the high-field satellite of H-4 (which resonates 12 Hz downfield from the low-field satellite of H-3), a negative signal for C-4 is



FIG. 2. (a) 500 MHz proton spectrum of a 16 mM solution of chrysene in CDCl₃, recorded using the decoupler coil of a 10 mm broadband probe on a NT-500 spectrometer. (b) Conventional protondecoupled ¹³C spectrum obtained from 200 scans (12 min). (c)-(g) SPT spectra obtained by transfer from the low-field ¹³C satellites of H-3, H-2, H-5, H-6, and H-1, respectively. Negative signals in those spectra result from SPT transfer of the high-field satellites of H-4 and H-5 which are also affected by the soft 180° pulses applied to the downfield satellites of H-3 and H-2. Each SPT spectrum is the result of 160 scans (10 min).

detected. Similarly, Fig. 2d shows the SPT spectrum from the lower field satellite of H-2 (positive resonance C-2), in which the selective ¹H pulse also affects the high-field satellite of H-5, and yields a negative signal for C-5. Figure 2e shows the SPT spectrum obtained by applying the selective 180° pulse 85 Hz downfield from the chemical shift of H-5, and clearly differentiates between C-5 (high intensity) and C-4 (low intensity). Finally, Figs. 2f and g show the transfer spectra for protons H-6 and H-1. Each SPT spectrum is the result of 160 accumulations (10 min).

The simplicity, high sensitivity, and high resolution of the decoupled SPT experiment make it very attractive for use in cases where only a limited number of ¹³C resonances have to be correlated. Of course, distinction between methyl, methylene, and methine resonances is easily possible by variation of the delay, Δ (16, 19). A drawback of the SPT experiment is the required knowledge of the approximate value of the ¹H-¹³C coupling constant. Since one-bond C-H couplings have been well studied (20) this usually does not present serious problems. The experiment is, of course, not limited to the detection of ¹³C, and can be applied directly to a number of other interesting nuclei.

ACKNOWLEDGMENTS

The authors thank Dr. E. D. Becker and Dr. J. A. Ferretti for many stimulating discussions and Laura Lerner for useful suggestions during the preparation of the manuscript.

REFERENCES

- 1. K. G. R. PACHLER AND P. L. WESSELS, J. Magn. Reson. 12, 337 (1973).
- 2. S. SØRENSEN, R. S. HANSEN, AND H. J. JAKOBSEN, J. Magn. Reson. 14, 243 (1974).
- 3. H. J. JAKOBSEN AND W. S. BREY, J. Am. Chem. Soc. 101, 774 (1978).
- 4. K. G. R. PACHLER AND P. L. WESSELS, J. Magn. Reson. 28, 53 (1977).
- 5. H. J. JAKOBSEN, S. A. LINDE, AND S. SØRENSEN, J. Magn. Reson. 15, 385 (1974).
- 6. A. A. MAUDSLEY AND R. R. ERNST, Chem. Phys. Lett. 50, 368 (1977).
- 7. G. BODENHAUSEN AND R. FREEMAN, J. Magn. Reson. 28, 471 (1977).
- 8. A. A. MAUDSLEY, L. MULLER, AND R. R. ERNST, J. Magn. Reson. 28, 463 (1977).
- 9. A. BAX, "Two-Dimensional Nuclear Magnetic Resonance in Liquids," Chap. 2, Reidel, Boston, 1982.
- 10. A. BAX AND G. A. MORRIS, J. Magn. Reson. 42, 501 (1981).
- 11. J. BRONDEAU AND D. CANET, J. Magn. Reson. 47, 419 (1982).
- 12. D. M. DODDRELL, W. BROOKS, J. FIELD, AND R. LYNDEN-BELL, J. Am. Chem. Soc. 105, 6973 (1983).
- 13. H. J. JAKOBSEN, H. BILDSOE, S. DONSTROP, AND O. W. SØRENSEN, J. Magn. Reson. 57, 324 (1984).
- 14. D. M. DODDRELL, W. BROOKS, J. FIELD, AND R. M. LYNDEN-BELL, J. Magn. Reson. 59, 384 (1984).
- 15. D. D. DAVIS, D. H. LIVE, W. C. AGOSTA, AND D. COWBURN, J. Magn. Reson. 53, 350 (1983).
- 16. D. P. BURUM AND R. R. ERNST, J. Magn. Reson. 39, 163 (1980).
- 17. H. J. JAKOBSEN, P. J. KANYHA, AND W. S. BREY, J. Magn. Reson. 54, 134 (1983).
- 18. A. BAX AND S. K. SARKAR, J. Magn. Reson. 60, 170 (1984).
- 19. D. M. DODDRELL AND D. T. PEGG, J. Am. Chem. Soc. 102, 6388 (1980).
- 20. P. E. HANSEN, Prog. Nucl. Magn. Reson. Spectrosc. 14, 175 (1981), and references therein.