

method reported here allows three of the four rings of this alkaloid target to be assembled with *complete* stereocontrol in only five steps from cyclopentanone and, moreover, directly introduces oxidation in the tricyclic product at the two desired sites. The structure of the crystalline dione **10** (mp 56 °C, from hexane) was confirmed by a single-crystal X-ray diffraction study.¹⁸

Since the products of acetal cyclization-pinacol rearrangements contain a ketone, the sequence reported here can be carried out in an iterative fashion to elaborate two new five-membered rings and accomplish a net two-carbon ring expansion of the starting ketone. The construction of the dicyclopentacyclooctane ring system, a tricyclic skeleton found in a number of biologically important sester- and diterpenes such as the fusicoccins and ophiobolins,¹⁹ illustrates this sequence. Hydroazulenone **6b**, readily available from cyclohexanone (see eq 2), was first elaborated by the efficient stereocontrolled sequence summarized in eq 3 to



* KHMDS, 0°C, THF; CH₂=CHCH₂I, -78° to -50°C, * OsO₄, NaIO₄, dioxane-H₂O (3:1), 23°C, ⁶ MeOH, TsOH (cat.), 23°C, ⁶ CH₂=CHLI (15 equiv), THF, -78° to 23°C, ⁶ Me₃SICH₂CO₂Et (20 equiv), Bu₄NF (cat.), 23°C, ¹ SnCl₄ (1.1 equiv), CH₂CH₂, -78° to 23°C.

provide 12 as a single diastereomer.²⁰ Rearrangement of 12 occurred smoothly in the presence of $SnCl_4$ to give the *cis*, *anti,cis*-dicyclopentacyclooctanones 13a and 13b in a 1:2 ratio and 59% yield after separation on silica gel. The most stable conformation of 13b, as determined by ¹H NMR NOE experiments and molecular mechanics calculations (MM2), is depicted in structure 14.

In summary, a wide variety of carbocyclic ring systems can be assembled efficiently and with excellent stereocontrol by the sequential acetal cyclization-pinacol rearrangement strategy reported here. This new chemistry significantly broadens the range of precursors potentially available for assembling carbocyclic skeleta since cyclopentane annulation is coupled with expansion of a preexisting ring. The studies described here, together with our earlier reports¹ and recent disclosures by Trost²² and Sworin,²³ clearly establish the utility of reaction designs that employ pinacol rearrangements to terminate cationic cyclizations.

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Supplementary Material Available: A typical procedure for the rearrangement step and experimental data for the X-ray diffraction study of 10 (7 pages). Ordering information is given on any current masthead page.

Three-Dimensional Heteronuclear NMR of ¹⁵N-Labeled Proteins

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The introduction of two-dimensional (2D) NMR¹ has made it possible to determine solution structures of small proteins.² Although the 2D approach greatly reduces spectral overlap, many ambiguities remain in the analysis of 2D protein NMR spectra because of coincident or nearly coincident chemical shifts. Commonly used procedures to solve this type of problem rely on the fact that the chemical shifts of many protons show different pH and temperature dependence. Another, more elegant approach utilizes 3D NMR³⁻⁶ to remove the problem of degenerate chemical shifts. Homonuclear 3D techniques, combining J connectivity and NOE information, have recently been demonstrated for small proteins, clearly demonstrating the power of this approach.^{6,7} However, for proteins larger than about 15 kD, the J connectivity transfer step in such a 3D experiment rapidly looses its efficiency, severely decreasing sensitivity. Here the use of a very sensitive 3D experiment is demonstrated for unraveling the regular protein NOESY spectrum. This method requires ¹⁵N labeling of the protein, a relatively simple procedure for bacterially overexpressed proteins. High-quality 3D NMR spectra can be obtained in a few days, without excessive demands for data processing or data storage.

The NOESY-HMQC pulse scheme we utilized (Figure 1) is slightly different from the scheme proposed very recently by Fesik and Zuiderweg,⁸ permitting observation of NOE's to $C\alpha$ H protons that resonate very close to the H₂O resonance. The t_1 and t_3 dimensions represent the time variables in a regular NOESY experiment; during the t_2 dimension the NH protons are labeled with their ¹⁵N chemical shifts. Therefore, a projection of the 3D spectrum onto the F_1, F_3 plane corresponds to the regular amide region of a 2D NOESY spectrum. However, individual F_1, F_3

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Figure 1. Pulse scheme of the 3D NOESY-HMQC experiment. The ¹H carrier is positioned in the center of the amide region, and the observe transmitter is used in the low power mode for generating a DANTE type presaturation of H₂O.¹⁴ The phase cycling used is as follows: $\phi = x, y$, $-x, -y; \psi = 4(x), 4(-x);$ Acq. = 2(x), 4(-x), 2(x), with data for odd- and even-numbered scans stored separately. The entire sequence is repeated with ψ incremented by 90° to obtain complex data in the F_2 dimension.

planes show NOE connectivities only for amide protons with a particular ¹⁵N chemical shift. Suppression of the intense H₂O resonance is accomplished in two stages; first the water resonance is slightly attenuated by the use of a very weak presaturating field (equivalent to $\gamma B_2 \approx 12$ Hz on the water resonance) and second an off-resonance $45^{\circ}-\tau-45^{\circ}$ pulse is used at the end of the NOESY mixing period to minimize the excitation of the remaining H₂O magnetization.

Figure 2 presents two (out of 64) NOESY slices taken from the most crowded region of the 3D spectrum at ¹⁵N shifts of 121.4 and 122.3 ppm. A slice taken at 121.8 ppm and the comparable region of a regular NOESY spectrum are shown in the supplementary data. The 3D spectrum has been recorded in 2.5 days on a NT-500 spectrometer, for a 1.8 mM solution of ¹⁵N labeled Staphylococcal nuclease (S. Nase) complexed with pdTp and Ca²⁺, in 90%H₂O/10\%D₂O, pH 6.5, 37 °C. The sensitivity of the 3D spectrum is similar to that of the conventional NOESY spectrum, but the severe overlap present in the 2D spectrum is almost completely removed. The 3D spectrum is particularly useful for tracing out the backbone (NH/NH and NH/C α H) connectivities in the protein and such correlations have been labeled in Figure 2. On the basis of this 3D spectrum and on a number of other isotopic labeling and double labeling techniques9-11 we have been





Figure 2. Two adjacent F_1/F_3 slices taken from the 500-MHz 3D spectrum of S. Nase, recorded with the scheme of Figure 1, for an NOE mixing time of 125 ms. The spectrum was recorded without using dummy scans; instead a single 90° pulse was used prior to the first scan of each (t_1, t_2) value to accomplish a "steady state". The ¹H and ¹⁵N carrier frequencies were positioned at 8.67 and 120.0 ppm downfield from the resonances of trimethylsilyl propionate and liquid ammonia, respectively. (A) Slice taken for an 15 N chemical shift (F₂) of 121.4 ppm and (B) for F₂ = 122.3 ppm. The spectrum results from a $32 \times 128 \times 256$ complex data matrix, with acquisition times of 23, 21, and 64 ms in the t_1 , t_2 , and t_3 dimension, respectively. 16 scans (twice the minimal number) were acquired per t_1/t_2 value and the total measuring time was 2.6 days. Zero filling was used in all three dimensions to yield a 64 × 256 × 512 matrix for the absorptive part of the 3D spectrum. Digital resolution in the final spectrum is 22, 24, and 8 Hz per point in the F1, F2, and F3 dimension, respectively. Digital filtering used was the following: 60° shifted sine bell (t3) and double shifted (60° at the beginning and 10° at the end) in the t_1 and t_2 dimensions. A linear base line correction routine was used in the F_3 dimension only. Data have been processed with commercially available software (New Methods Research Inc., Syracuse, NY) supplemented by simple home-written routines for digital filtering and Fourier transformation¹³ in the t_2 dimension.

Major practical problems with 3D spectroscopy are the long measuring time needed to get sufficient digitization in the F_1 and F_2 dimensions of the 3D spectrum and the large size of the 3D matrix. We have used folding of some of the resonances in both the F_1 and F_2 dimensions and employed an unfolding procedure¹² in the F_1 dimension (based on shifting the F_1 carrier position during data processing¹³) to obtain maximum resolution with a relatively small number of t_1 and t_2 increments. The minimum measuring time also depends on the number of scans needed for phase cycling or each set of t_1, t_2 values and on the overhead time needed to write the data to disk at the end of an acquisition. In principle, replacing the first ¹H 90° pulse by a frequency-selective pulse could reduce the minimum measuring time (or increase digitization),³ but this would eliminate informative correlations from the 3D spectrum. We therefore believe that the heteronuclear 3D experiment discussed here is best executed in a nonselective fashion. The sensitivity of the heteronuclear 3D technique is excellent, and resonance overlap in the S. Nase 3D spectrum is minimal, despite the relatively coarse digitization. The three-dimensional NMR experiment reported here should be applicable to proteins significantly larger than S. Nase.

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Supplementary Material Available: Three figures giving the projection of the 3D spectrum on the F_2/F_3 plane, a F_1/F_3 slice taken at $F_2 = 121.8$ ppm, and an identical region of the regular 2D NOESY spectrum (4 pages). Ordering information is given on any current masthead page.

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A General Solution To Implementing the 4π Participation of 1-Aza-1,3-butadienes in Diels-Alder **Reactions: Inverse Electron Demand Diels-Alder** Reactions of α,β -Unsaturated N-Benzenesulfonyl Imines

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The Diels-Alder 4π participation of simple α,β -unsaturated imines is rarely observed and typically suffers low conversions, competitive imine addition, and/or imine tautomerization precluding [4 + 2] cycloaddition.² Consequently only a limited number of 1-aza-1,3-butadiene structural variations and modified or restricted reaction conditions have been introduced that have permitted the productive 4π participation of selected α,β -unsaturated imines in [4 + 2] cycloaddition reactions.³⁻⁷ In the conduct

Scheme I



of synthetic efforts on natural and synthetic quinoline-5,8-quinones including streptonigrone,8 we have examined alternative approaches to predictably control and accelerate the intermolecular 4π participation of 1-aza-1,3-butadienes in [4 + 2] cycloaddition reactions. The complementary N-1 or C-3 substitution of an α , β -unsaturated imine with an electron-withdrawing substituent would be expected to accentuate the electron-deficient nature of the 1-aza-1,3-butadiene and accelerate its potential [4 + 2] cycloaddition reaction with electron-rich dienophiles in LUMO_{diene}-controlled Diels-Alder reactions.² In addition, a bulky, electron-withdrawing N-1 1-aza-1,3-butadiene substituent would be expected to preferentially decelerate 1,2-imine addition relative to [4 + 2] cycloaddition and convey [4 + 2] cycloaddition product stability to the reaction conditions while enhancing the electron-deficient nature of the diene. Herein we detail a comparative study of the 4π participation of N¹-substituted α,β -unsaturated imines in LUMO_{diene}-controlled Diels-Alder reactions which has revealed the general, well-defined 4π participation of α,β -unsaturated N-benzenesulfonyl imines in regio- and endo-specific inverse electron demand Diels-Alder reactions suitable for the diastereoselective preparation of substituted N-benzenesulfonyl-1,2,3,4-tetrahydropyridines.

Representative results of initial studies employing stable imine derivatives of 1-acetylcyclohexene9 are summarized in eq 1 and

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